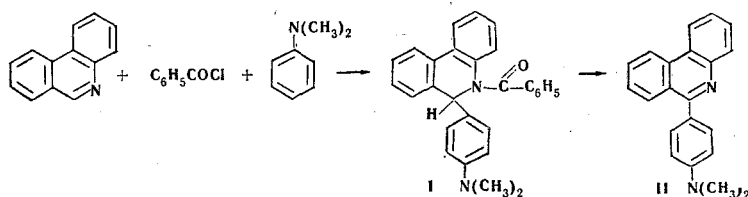


INTRODUCTION OF THE PHENANTHRIDINE RESIDUE
INTO NUCLEOPHILIC ORGANIC COMPOUNDSA. K. Sheinkman, A. P. Kucherenko,
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UDC 547.836.3

Existing methods for the preparation of 6-substituted phenanthridines from o-aminobiphenyl are inconvenient as a result of the inaccessibility of the latter. We have developed a very simple method for the preparation of these derivatives by reaction of phenanthridine, which is an unsaleable by-product of the coking process, with nucleophilic organic compounds, in the presence of acyl halides.

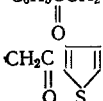
Thus, on heating phenanthridine with dimethylaniline and benzoyl chloride, 5-benzoyl-6-p-dimethylaminophenyl-5,6-dihydrophenanthridine (I) was obtained in high yield. Alkaline hydrolysis of this afforded the known 6-p-dimethylaminophenylphenanthridine (II) [1].



Similarly, the phenanthridine residue was readily introduced into the activated aromatic rings of 1-alkyl-1,2,3,4-tetrahydroquinolines, 1-alkyl-2,3-dihydroindoles, and into the molecules of nucleophilic five-membered heterocycles and ketones.

The yields and properties of the compounds obtained are given in Table 1.

TABLE 1

R	Mp, °C (solvent)	Molecular formula	Found, %			Calculated, %			Yield, %
			C	H	N	C	H	N	
<i>p</i> -C ₆ H ₄ N(CH ₃) ₂	184—185 C ₂ H OH: petro- leum ether, 5:1)	C ₂₈ H ₂₄ N ₂ O	82,92	6,06	6,74	83,13	5,98	6,93	65
1-CH ₃ -1,2,3,4- tetrahydro-6- quinolyl	184—186 isoamyl alcohol	C ₃₀ H ₂₆ N ₂ O	83,71	6,58	6,31	83,69	6,09	6,50	60
1-CH ₃ -2,3-dihydro- 5-indolyl	115—116 amyl alcohol	C ₂₉ H ₂₄ N ₂ O	83,19	5,93	6,71	83,62	5,81	6,72	62
C ₆ H ₅ CCH ₂	149—150 ethanol	C ₂₈ H ₂₁ NO ₂	82,96	5,26	3,96	83,35	5,25	3,47	17
	209—210 ethanol	C ₂₆ H ₁₉ NO ₂ S	76,4	5,0	3,50	76,26	4,67	3,43	15

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